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chemistry, bases are substances
that, in aqueous solution, release
hydroxide (OH⁻) ions, are
slippery to the touch, can taste
bitter if an alkali, change the
color of indicators (e.g., turn red
litmus paper blue), react with
acids to form salts, promote
certain chemical reactions (base
catalysis), accept protons from
any proton donor, and ...
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polar inorganic compound that is
at room temperature a tasteless
and odorless liquid, which is
nearly colorless apart from an
inherent hint of blue.It is by far
the most studied chemical
compound and is described as the
"universal solvent" and the
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hydroxide is a highly caustic
substance that is used to
neutralize acids and make sodium
salts. (From Merck Index, 11th
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oil with a long and interesting
history. A staple of Indian,
Arabic, and North African
cooking, it is also found
throughout Indonesian societies.
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lÃ¡mites. De manera menos
general y precisa, se ha definido
molÃ©cula como la parte mÃ¡s
pequeÃ±a de una sustancia
quÃ©mica que conserva sus
propiedades quÃ©micas, y a partir
de la cual se puede reconstituir la
sustancia sin reacciones
quÃ©micas. MolÃ©cula -
Wikipedia, la enciclopedia libre -
In chimica e in fisica, la molecola
(dal latino scientifico molecula,
derivato a sua volta da moles, che
significa "mole", cioÃ¨ "piccola
quantitÃ") Ã un'entitÃ
elettricamente neutra composta da

due o piÃ¹ atomi, dello stesso
elemento o di elementi diversi,
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